

Conversion of HI Molecule to H₂ Molecule

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Abstract: *In this article I review the historical development and conversion of atomic to molecular hydrogen in astronomy. I discuss how the discoveries of HI and H₂ in the interstellar medium were followed by studies of the relative abundance of atomic and molecular gas. Understanding this led to increasingly sophisticated theoretical models for H₂ formation on the surface of interstellar dust grains. In certain situations, astronomical data can be used to constrain the formation rate of H₂ molecules. Finally, I use the reasonably well-determined chemistry of HI and H₂ to determine the overall timescale of star formation.*

Keywords: HI Molecule, H₂ Molecule, interstellar dust grain, time scale, hyperfine transition, theoretical model

1. INTRODUCTION :

Neutral atomic hydrogen is a normal, electrically neutral hydrogen atom with one proton and one electron. It is commonly referred to as HI (pronounced H-one), and is located throughout galaxies as HI clouds or external to galaxies as part of the inter cloud gas. It is detected via the spin-flip transition at 21cm in the radio, and HI clouds were used to determine the structure of our Galaxy from our location within it. Atomic hydrogen (HI) is the most abundant atom in the universe and H₂ the most abundant molecule. The importance of these two species and their obvious inter-connection has led to an ongoing interest in their relationship in a wide variety of astrophysical situations. My focus here is on the formation of H₂ in the context of evolution of material from more diffuse, atomic regions, to denser, primarily molecular clouds of gas, which is the initiation of the process of star formation.

2. DETECTION OF ATOMIC AND MOLECULAR HYDROGEN IN THE INTERSTELLAR MEDIUM

The possibility of observing HI through its 21 cm hyperfine transition was predicted by van der Hulst (1945) and by Shklovski (1948). These papers were strong motivations for experimental searches. Different groups had near-simultaneous success a few years later (Ewen & Purcell 1951, Muller & Oort 1951, and report by Pawsey 1951). The work of the Harvard group was the extraordinary PhD research of H. Ewen, who pursued this effort despite his expectation that

only an upper limit would result (Ewen 2007). The initial detection of HI was very quickly followed up by large-scale studies of HI in the Milky Way, and the value of 21cm astronomy was quickly established.

Initial work suggested that dust and atomic hydrogen densities are proportional even in clouds with modest visual extinction ($T_{\text{dust}} \leq 3$; Lilley 1955). This correlation was, however, found to break by Bok, Lawrence, & Menon (1955). In a very early study of the outer galaxy, van de Hulst, Muller & Oort (1954) found only 30% of the HI intensity in Taurus expected from dust obscuration, and attributed this to conversion of hydrogen into molecular form. In a more detailed study of the same region, Garzoli & Varsavsky (1966) found that there was an inverse correlation between HI column density and visual extinction. Mészáros (1968) found that the column density of atomic hydrogen in the dust cloud near ρ Ophiuchi was essentially independent of dust column density and possibly decreased along lines of sight having the greatest column density of dust. He also suggested that the hydrogen is being converted to molecular form in this dense regions.

Following the remarkably early discussions about the formation and importance of H₂ (van de Hulst 1948 and retrospective discussion in van de Hulst 1997), sophisticated theories of H₂ formation on grains were developed in the 1960's (McCrea & McNally 1963, Gould & Salpeter 1963). These undoubtedly encouraged the search for H₂ which is, however, extremely difficult to observe in the general

interstellar medium due to lack of energy levels with appreciable population at low temperatures, and the weak transitions between rotational levels of the homonuclear molecule. The challenging experimental prospect and the importance of H₂ were both discussed in detail by Field, Somerville, & Dressler (1966).

H₂ was first detected by Carruthers (1970), after an earlier unsuccessful attempt (Carruthers 1967; see Carruthers 1969 for a description for equipment). These initial rocket observations showed that towards the star ζ Per the atomic and molecular column densities were approximately $4 \times 10^{20} \text{ cm}^{-2}$. This work was soon followed by Copernicus satellite studies showing that a significant fraction of the interstellar gas in the direction of many stars with color excess $\geq 0.1 \text{ mag}$ was molecular, while only a very small fraction of the gas was molecular for unreddened stars (Spitzer et al. 1973).

Tracing H₂ directly through UV absorption in regions of greater column density is difficult because the dust extinction at UV wavelengths results in severe attenuation of background sources. Infrared rotation-vibration lines have been utilized to a limited extent (Lacy et al. 1994). The pure rotational lines of H₂ are at longer wavelengths (up to $\lambda=28 \mu\text{m}$), but they still require temperatures of over 100 K to be seen in emission. This means that in a general interstellar medium they are not detected. Results have been obtained on selected nearby regions with higher temperatures (e.g. Neufeld et al. 2006) and surprisingly strong emission has been detected from certain galaxies (Applenton et al. 2006).

In general, astronomers assume that the H₂ in well-shielded regions having $A_V = 1 \text{ mag}$ is entirely molecular, and typically reference abundances of other molecular species to an inferred, albeit unobserved column density of H₂.

3. THEORETICAL MODELS OF H₂ FORMATION

Many experts in this field have attended to present the state of the art in theoretical calculations for

formation of H₂ on the surfaces of dust grains. I here only broadly review the historical development of this subject. I shall not discuss H₂ formation by purely gas phase reactions. For formation of H₂ at low densities in the early universe, reactions involving H₂⁺ and H₂ are dominant (Galli & Palla 1998). At higher densities, three body reactions become more important (Palla et al. 1983), and these are critical for production of H₂ during the formation of the first stars in the universe (Yoshida et al. 2006).

As mentioned above, detailed theoretical modeling of H₂ formation on dust grains dates back to at least the 1960's with the paper by McCrea & McNally (1963), which even included a grain size distribution. The approach adopted was to use a rate equation to describe the changes in the average properties of a volume of an interstellar cloud. It was assumed that an H atom in the gas phase sticks to a grain which it hits. It is also assumed that there is at least one H atom on the surface of a grain, so that when a hydrogen atom hits it and sticks, the two atoms will recombine to form a H₂ molecule, with sufficient energy released to desorb the newly-formed molecule from the grain. The number of H₂ molecules formed per unit time per unit volume of the cloud is proportional to the number density of hydrogen atoms, $n(\text{H})$ and of grains, n_g . However, the number of grains is proportional to the total proton density n_0 . If we define $n_1 = n(\text{H})$, $n_0 = n_1 + 2n_2$, and we can write

$$\frac{dn_2}{dt} = F n_1 n_0, \quad (3.1)$$

where F is the formation rate constant, having units $\text{cm}^3 \text{ s}^{-1}$. It includes a number of factors, including the probability that the incident atom will stick to the grain, and the grain area per proton in the gas. In the simplest approximation, the only temperature dependence is due to the speed of the H atoms, and taking $T=100 \text{ K}$, McCrea and McNally derived $F=7 \times 10^{-17} \text{ cm}^3 \text{ s}^{-1}$.

The physics of the dust grain surface was analyzed in considerably more detail by Hollenbach & Salpeter

(1971). These authors considered physisorption (attraction by Van der Waals forces) holding H atoms on the grain, as well as the atomic mobility, which is essential for two adsorbed atoms to be able to interact and form a H₂ molecule. They wrote the formation rate of H₂ in a more detailed form

$$\frac{dn_{H_2}}{dt} = \frac{1}{2} \langle v_H \rangle n_H n_1 \langle \sigma g \rangle, \quad (3.2)$$

where γ is the sticking probability for incident H atoms, $\langle v_H \rangle$ is their mean thermal velocity, and $\langle \sigma g \rangle$ is the average grain cross section.

This issue had been addressed earlier by Knapp et al. (1966), who pointed out that if there were only physisorption, temperatures above 8 K would result in the probability of having two atoms simultaneously on the grain being so low that efficient recombination would be impossible. These authors also indicated that chemical adsorption (resulting from formation of a chemical bond between the H atom and an atom on the grain surface) would improve the prospect for efficient H₂ formation over a wide range of temperatures. Hollenbach & Salpeter also suggested that a “semi-chemical” enhancement of the binding energy of the H atoms could increase the temperature range for efficient H₂ formation.

Cazaux & Tielens (2004) developed a grain model including both sites that can physisorb H atoms and those that can provide chemisorption. While the specific binding energies and the barrier height separating the two types of sites, are uncertain, this model has the welcome effect of broadening the temperature range of efficient H₂ formation to cover from somewhat below 10 K to somewhat above 30 K. While still not encompassing all regions in which efficient H₂ formation is indicated, this does at least include dark clouds and the majority of giant molecular cloud material. Chang, Cuppen, & Herbst (2005) studied “mixed” grain surfaces composed of carbon and olivine material, and found that the temperature range for efficient H₂ formation was moderately increased over that for homogeneous surfaces, for

which it was very small. An investigation of rough grains by Cuppen & Herbst (2005) indicates that H₂ formation can proceed at temperatures up to 50 K, with this limit depending on the details of bonding to the non-planar surface structure. More realistic models in this vein are obviously called for as we learn more about the surface properties of real interstellar grains.

As indicated in eq. 3.2, the rate of H₂ formation is also dependent on the sticking probability. This has received considerable attention over the years, and certainly depends on grain composition and morphology. Buch and Zhang (1991) calculated the sticking probability of H and D atoms on amorphous ice particles, finding it to be close to unity at temperatures of dense interstellar clouds ($T \leq 20$ K), but dropping at higher temperatures. This may not be representative of all grains types and sizes but gives some confidence that γ is in general of order unity.

Using a rate equation to describe the evolution of cloud constituents has definite limitations in that it ignores the discrete nature of the atoms and molecules involved. While it may be valid in the limit of large atomic fluxes and atoms on a grain, it is evident that in the situation with very little atomic hydrogen in a cloud, there can be less than one hydrogen atom absorbed on a grain, so that it is not correct to assume that the rate derived for a much higher population of H atoms applies. Considerations of this sort have motivated a large number of investigators to develop models that deal with the discrete nature of particles more explicitly. These generally are more complex than rate equation models, and are difficult to describe in any convenient manner. A good review of this topic is given by Herbst, Chang, & Cuppen (2005) which also has an extensive list of references.

4. THE USE OF HI AND H₂ TO DETERMINE TIME SCALES IN INTERSTELLAR CLOUDS

As early as in 1968, it was suggested that the time-dependent conversion of atomic to molecular hydrogen could be used to determine an “age” for

an interstellar cloud; Mészáros (1968) took this as the time since the cloud was last ionized by a hot star. Solomon & Werner (1971) and Shu (1973) addressed the question of how the amount of HI in a “molecular” cloud could be used to constrain its evolutionary history.

The UV measurements referred to in sec.2 measure atomic and molecular column densities along lines of sight towards hot stars with modest foreground extinctions (few mag.). To probe the atomic to molecular hydrogen ratio in regions of greater extinction is possible in principle using the pure rotational transitions of H₂, but these are so weak that the spectral features are not generally detectable with presently-available equipment. We are left with using surrogates to trace the column density of H₂; these are generally isotopologues of carbon monoxide which are optically thin, namely ¹³CO and O.

For tracing the atomic gas, the situation is challenging because 21 cm line emission is so widespread. Observations detect emission with large line widths in almost all directions, which makes it difficult to associate a specific velocity feature with a certain region of space along the line of sight. One situation which is more favorable is looking for HI self-absorption. Such features in 21 cm spectra are common, and can result from temperature variations as a function of velocity (different clouds of gas), or from variations along the line of sight. One situation that has proven very useful is when we detect a very narrow absorption feature in the HI spectrum. The very narrow absorption features are found to be highly correlated in terms of nonthermal line width and spatial extent with optically thin carbon monoxide isotopologues (Li & Goldsmith 2003; Goldsmith & Li 2005). This facts together with the line widths themselves indicating very low temperatures, sometimes below 20 K, suggest that the atomic gas producing the HI narrow self-absorption (or HISA) features is mixed with the molecular gas in the well-shielded central regions of dense molecular clouds.

5. SUMMARY

The issue of HI to H₂ conversion remains a critical one for astrophysics. Laboratory measurements, theoretical models, and astronomical data are being used together to understand the complex surface chemistry and radiative transfer issues involved. I can look forward to continued progress in all of these areas, which will allow me to use the atomic to molecular hydrogen abundance ratio as an importance diagnostic of the evolution of the interstellar medium.

REFERENCES

1. Appleton, P.N., Xu, K.C., Reach, W., Dopita, M.A., Gao, Y., Lu, N., Popescu, C.C., Sulentic, J.W., Tuffs, R.J., & M.S. 2006, ApJ, 639, L51
2. Bok, B.J., Lawrence, R.S., & Menon, T.K. 1955, Pub. Astr. Soc. Pacific, 67, 108
3. Buch, V. & Zhang, Q. 1991, ApJ, 379, 647
4. Crruthers, G.R. 1967, ApJ, 148, L141
5. Carruthers, G.R. 1969, Appl. Optics, 8(3), 633
6. Crruthers, G.R. 1970, ApJ, 161, L81
7. Cazaux, S. & Tielens, A.G.G.M 2004, ApJ, 604, 222
8. Chang, Q., Cuppen, H.M., & Herbst, E. 2005, A&A, 434, 599
9. Chang, Q., Cuppen, H.M., & Herbst, E. 2006, A&A, 458, 497
10. Ewen, H. & Purcell E.M. 1951, Nature, 169, 356
11. Ewen, H. 2007, private communication
12. Field, G.H., Somerville, W.B., & Dressler, K. 1966, Ann. Rev. Astr. Astrophys., 5, 207
13. Galli, D. & Palla, F. 1998, A&A, 335, 403
14. Garzoli, S. L. & Varavslly 1966, ApJ, 145, 79
15. Gould, R.J. & Salpeter, E.E. 1963, ApJ, 138,393
16. Herbst, E., Chang, Q., & Cuppen, H.M. 2005, Chemistry an Interstellar Grains, in Journal of Physics Conference Series 6 (2005) Light, Dust and Chemical Evolution (:IOP Publishing), 18
17. Hollenbach, D. & Salpeter, E.E. 1971, ApJ, 163, 155
18. Knapp, H.F.P., van den Meijdenberg, C.J.N., Beenakker, J.J.M, & van de Hulst, H.C. 1966, B.A.N. 18, 256

19. Li, D. & Goldsmith, P.F. 2003, ApJ, 585, 823
20. Lilley, A.E. 1955, ApJ, 121, 559
21. McCrea, W.H. & McNally, D. 1960, MNRAS, 121, 238
22. Mészáros, P. 1968, Astr.Sp.Sci., 2, 510
23. Muller, C.A. & Oort, J.H. 1951, MNRAS, 121, 238
24. Neufeld, D.A., Melnick, G.J., Sonnentrucker, P., Bergin, E.A., Green, J.D., Kim, K.H., Watson, D.M., Forrest, W.J., & Pipher, J.L. 2006, ApJ, 649, 816
25. Palla, F., Salpeter, E.E., & Sathler, S.W. 1983, ApJ, 271, 632
26. Pawsey, J.L. 1951, Nature, 168, 357
27. Shklovski, I.S. 1948, Astron. Zh. SSSR, 26(1948)1, 10
28. Shu, F.H. On the Genetic Relation Between Interstellar Clouds and Dust Clouds, in Interstellar Dust and Related Topics, ed. J.M. Greenberg & H.C. van de Hulst (Dordrecht:Reidel), 257
29. Soloman, P.M. & Wrener, M.W. 1971, ApJ, 165, 49 257
30. Spitzer, L., Drake, J.F., Jenkins, E.B., Morton, D.C., Rogerson, J.B., & York, D.G. 1973, ApJ, 181, L116
31. van de Hulst, H.C. 1945, Ned. Tijds. Natuurkunde, 210
32. van de Hulst, H.C. 1948, Harvard Observatory Monographs, Centennial Symposia Dec. 1946, 73
33. van de Hulst, H.C., Muller, C.A., & Oort, J.H. 1954, B.A.N., XII(452), 117
34. van de Hulst, H.C. 1997, Molecules in Astrophysics Half a Century Ago, in Molecules in Astrophysics: Probes and Processes, ed. E. van Dishoeck (Dordrecht: Kluwer), 13
35. Yoshida, N. Omukai, K., Hernquist, L., & Abel, T. 2006, ApJ, 652, 6