

Enhanced electronic and magnetic performance of MoS₂ monolayer via Tc & Nb impurities defect and water adsorption on impurities defected materials

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Abstract

This study examined the effect of Tc & Nb impurity atoms on MoS₂ (Tc-MoS₂ & Nb-MoS₂), and adsorption of water molecule on impurities defected MoS₂ (Tc-W-MoS₂ & Nb-W-MoS₂) material from first-principles calculations. By the estimation of their ground state energy and binding energy, they are stable 2D materials. From band structure and density of states (DoS) calculations, Tc & Nb impurities affect the nature of pristine MoS₂. It is found that Tc-MoS₂ has n-type & Nb-MoS₂ has p-type semiconducting nature. Water interaction on Tc-MoS₂ & Nb-MoS₂ slightly changes the electronic properties and impacts the bandgap, which enhanced the electronic performance of material than that of pristine MoS₂. The magnetic properties of Tc-MoS₂, Nb-MoS₂, Tc-W-MoS₂, and Nb-W-MoS₂ are analyzed and found to exhibit an uneven distribution of up-spin and down-spin states of electrons in the orbital of atoms near the Fermi level. It reflects that they have magnetic properties. The non-magnetic MoS₂ material changes in to weak magnetic defected-MoS₂ materials due to the presence of Tc, Nb and adsorbed water molecule. It means, impurity defects add to magnetic properties of pristine MoS₂. Magnetic properties on defected MoS₂ occurred due to the dominant contributions of spin states of 4d-orbital of Mo, Tc, Nb atoms, and 3p-orbital of S atoms in the structures. This study highlights the impact of Tc & Nb impurity atoms and adsorbed water molecule on impurities defected MoS₂. The studied materials have potential applications in the fields of catalysis, nanoelectronics, biomedicine, and magnetic sensors on the basis of their electronic and magnetic properties.

Keywords

Adsorption, Defect, Fermi, Impurity, semiconducting.

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1 Introduction

Two-dimensional (2D) transition-metal dichalcogenides (TMDCs) are a type of compound that consists of a transition metal from group IV-X (e.g., Mo or W) and a chalcogen element (S, Se, or Te) arranged in the form MX_2 [1]. These compounds can have a variety of properties, including metallic (such as VS_2 and NbS_2), semiconducting (such as MoS_2 and WS_2), or insulating (like HfS_2) [2, 3]. TMDCs have a structure where one layer of M atoms is sandwiched between two layers of X atoms, and these layers are held together by van der Waals (vdWs) force. Their structure is composed of a honeycomb pattern [4, 5] in which the three atomic planes (chalcogen-metal-chalcogen) stack to create the individual layers of the material [6]. The MoS_2 monolayer is a semiconducting TMDC material that has Mo and S atoms stacked together in an S-Mo-S configuration, which results in a direct bandgap of 1.80 eV [5–9]. The unique properties of hexagonal TMDC materials include their atomic-scale thickness as well as their direct bandgap in its electronic band (wide bandgap exotic semiconductor), which can vary depending on the stacking order [3, 7, 10]. The weak interlayer forces, strong spin-orbit coupling, and robustness of MoS_2 make it an attractive material for electronic and mechanical applications due to its layered structure [11–13]. Due to its high on/off current ratio and higher carrier mobility at room temperature, MoS_2 has applications in field-effect transistors (FETs) and photodetectors [14, 15]. 2D semiconductors, both in the form of monolayer and multilayer, have a wide range of applications. They include usage in transistors, signal amplifiers and integrated logic circuits [16]. They also have potential uses in photocatalysts, solar cells, spintronics, flexible optoelectronics, nanoelectronics, nanophotonic, nanosensing, energy harvesting, photovoltaic solar cells, photocatalytic cells [17–19], DNA sequencing, biological and chemical sensors, lubricants, and catalytic surfaces for hydrogen storage in technology and industrial sectors [20, 21].

The electronic properties of 2D materials can be affected by external factors such as strain, electric field, pressure, temperature, doping, and defects. This means that the properties of TMDCs can be adjusted by controlling the environment they are in [22–25]. It is interesting to study how 2D materials like TMDCs interact with various solvents, such as water, due to their atomically flat surfaces which allow for different hydrophobic and long-range interactions [26, 27]. The areas around the edges and vacancies in 2D materials are highly sensitive to the adsorption of molecules, which can change the electronic and magnetic properties of the layers by distorting their arrangement [24, 28].

The way that H_2O interacts with MoS_2 can vary depending on where it is adsorbed [29]. The way that 2D materials interact with other materials, biomolecules, solvents and ions is affected by the structure of water at the interface [30–32]. The ability to tune the electronic structure of 2D-TMDs is made possible by the strong relationship between defects and chemical dopants in these materials, as well as the stacking of multiple monolayers [33, 34]. MoS_2 based devices are used in moisture environment too, so MoS_2 -based devices can also be used in moist environments. However, the adsorption of water molecules on the monolayer MoS_2 can lead to water-induced oxidation and the formation of Molybdenum Trioxide (MoO_3), which causes the loss of lubricity [35, 36]. Various studies have revealed that the accumulation of hydrocarbon contaminants in the air can cause MoS_2 to become more hydrophobic over time, and this can affect the material's electrical or structural properties [37]. Despite the amount of research on the interaction of MoS_2 with water, the interactions of impurity-defected MoS_2 with water have not been thoroughly examined. Understanding these interactions could lead to the development of more efficient 2D materials-based biological and chemical sensors for medical devices [38].

Crystalline structures inevitably contain defects, or deviations in the arrangement of atoms or ions, and are considered a necessary consequence of entropy considerations in solids [39, 40]. Imperfections in solid materials, such as the replacement of an atom with a foreign atom or the removal of an atom from the structure, are known as impurity and vacancy defects respectively. These defects play a crucial role in altering and utilizing the undesired properties of the material [41]. These defects can lead to the discovery of new electronic, magnetic, mechanical, and transport properties in materials. This can have potential applications in various fields including catalysis, nanoelectronics, biomedicine, magnetic sensors, and more. It is crucial to study the properties of 2D materials with defects and how they are affected by these imperfections [42, 43].

Previous studies have investigated the structural, electronic and magnetic characteristics of a pure MoS_2 monolayer [8, 9], but the effect of impurities such as Tc, Nb on MoS_2 and water adsorbed on impurities defected MoS_2 have not been reported. The current study aims to examine the impact of the impurity atoms on MoS_2 using first-principles calculations that combine spin-polarized density functional theory (DFT) method. The remainder of the paper is structured as follows: the methodology and computational details used in the study are discussed in Section 2, the findings, results and discussion are presented in Section 3, and

the final conclusions of the work are outlined in Section 4.

2 Methods and Materials

The study carried out structural optimization, electronic and magnetic properties calculations using density functional theory (DFT) method [44] implemented in the Quantum ESPRESSO code [45] using ultra-soft pseudopotentials (USPPs). The Grimme model of USPP suggests the use of the Rappe-Rabe-Kaxiras Joannopoulos (RRKJ) model to describe weak van der Waals (vdWs) interactions in the system. This model of USPPs includes only the chemically active valence electrons in the calculations, which reduces the complexity of the effects caused by the motion of non-valence electrons of an atom and its nucleus [46]. The valence electronic configuration in H, C, Mo, O, S, Tc and Nb atoms of our system are H: $1s^2$; O: [He] $2s^2 2p^4$; S: [Ne] $3s^2 3p^4$; Mo: [Kr] $4d^5 5s^1$; Tc: [Kr] $4d^5 5s^2$ and Nb: [Kr] $4d^5 s^1$ respectively. We used a (3×3) supercell structure of monolayer MoS₂ to perform our calculations. First, we created the unit cell of MoS₂ using the XCrySDen (structure visualization software) and Quantum ESPRESSO (QE) computational tools. We have used an experimental lattice parameter of value 3.19 Å in the input file. We then converged the kinetic energy cut-off value, k-points, and lattice parameters. We then constructed impurity defects in MoS₂ by replacing the central Mo atom with Nb and Tc respectively and then water molecule adsorbed on impurity defected MoS₂ are shown in Fig. 1(a-d) respectively. The study also used a plane wave basis set and generalized gradient approximation (GGA) with Perdew, Burke, and Ernzer (PBE) to incorporate electronic exchange and correlation (XC) potential in the calculations [47]. The plane wave basis set is used to implement kinetic energy cutoff and charge density cutoff in the calculations. The kinetic energy cutoff used is 35 Ry and the charge density cutoff is 10 times the kinetic energy cutoff, which is 350 Ry, for all systems using USPPs for plane wave expansion. A $(8 \times 8 \times 1)$ k-mesh generated using the Monkhorst-Pack (M-P) scheme is used to integrate the Brillouin Zone (BZ) during self-consistent calculations and structure optimization in the first irreducible Brillouin Zone [48, 49]. The calculations are stopped when the total energy and force reached a threshold of 10⁻⁴ Ry and 10⁻³ Ry/Bohrs respectively to ensure accurate results. The optimization of all structures is done using the Broyden-Fletcher-Goldfarb-Shanno (BFGS) [50] algorithm until the change in total energy is less than 10⁻⁴ Ry and the force is less than 10⁻³ Ry/Bohrs between two consecutive scf iterations. The calculations used a small broadening width of 0.001 Ry using the Marzari-Vanderbilt (M-V) smearing

method to ensure the accuracy of the results [51]. The "david" diagonalization method with a mixing factor of 0.6 is chosen for self-consistency. In order to study magnetic properties, spin-polarized density of states (DoS) and partial density of states (PDoS) calculations are performed. The systems are relaxed by using optimized values of kinetic energy cutoff, charge density cutoff, k-points, and lattice parameters. The self-consistent total energy calculations are done after the relaxation. The band calculations are done by selecting 100 k-points along the high symmetry points in the reciprocal lattice. These k-points are the sampling points in the first Brillouin Zone of the reciprocal lattice. Before calculating the density of states (DoS) and partial density of states (PDoS), non-self-consistent (nscf) calculations are done using an automatic denser mesh of $(16 \times 16 \times 1)$ k-points, with a verbosity setting of "high" and an occupation setting of "tetrahedra".

3 Results and Discussion

In this section, we discussed in detail about the structural, electronic, and magnetic properties of pristine MoS₂, Tc & Nb impurities defected MoS₂ materials, and water adsorption on impurities defected MoS₂ materials. The analysis is based on spin-polarized DFT method of calculations.

3.1 Structural Analysis

In this section, we have discussed the structural properties of pristine MoS₂, Nb & Tc impurities defected MoS₂, and water adsorbed on impurities defected MoS₂ materials. Firstly, we have prepared unit cell of MoS₂ and check its stability by calculating its ground state energy and binding energy, it is found to be stable 2D material. After that, we have prepared (3×3) supercell structure of MoS₂ by extending unit cell along x-and y-axis, which is used for further calculations. So, we examined the stability of (3×3) supercell structure of MoS₂ by estimating its ground state energy -1741.61 Ry and binding energy 6.83 eV, and found that it is stable material. This estimated binding energy value is comparable with reported value of other stable 2D materials [52, 53]. The binding energy of MoS₂ is found by using equation (1) [24];

$$(E_b)_{Mo-S_2} = [E_{MoS_2} - N E_{Mo} - N' E_S] / N'' \quad (1)$$

where, $(E_b)_{Mo-S_2}$ is the binding energy per MoS₂ pair, E_{MoS_2} , E_{Mo} & E_S are the ground state energy of MoS₂, single Mo & single S atoms respectively, N & N' respectively represent the number of Mo atom & S atom present in supercell structure, N'' indicates the total number of atoms present in supercell structure. Moreover, we have constructed Tc impurity defected MoS₂ (Tc-MoS₂) material by

replacing Mo sites atoms from stable (3×3) supercell structure of MoS₂, and then relax calculations are done. The minimum ground state energy of value -1774.07 Ry and maximum binding energy of value 7.32 eV are obtained at the center position of Mo atom of Tc-MoS₂, is shown in Fig. 1(a). This value is close with reported value of other 2D materials [52, 53]. Similar procedure is done for the fabrication of Nb impurity defected MoS₂ supercell structure (Nb-MoS₂). The estimated ground state energy and binding energy of Nb-MoS₂ are found to be -1711.97 Ry & 6.42 eV respectively. The estimated binding energy of Nb-MoS₂ agrees with reported value of other 2D materials [52, 53]. Hence, Nb-MoS₂ is a stable 2D material, is shown in Fig. 1(b). The binding energy of Tc-MoS₂ and Nb-MoS₂ are obtained by using equation (2) [24];

$$(E_b)_{Tc/Nb-Mo-S_2} =$$

$$\frac{[E_{Tc/Nb-MoS_2} - NE_{Mo} - N'E_S - N''E_{Tc/Nb}]}{N'''} \quad (2)$$

where, $(E_b)_{Tc/Nb-Mo-S_2}$ is the binding energy per Tc-MoS₂ pair, $E_{Tc/Nb-MoS_2}$, E_{Mo} , $E_{Tc/Nb}$ & E_S are the ground state energy of Tc/Nb-MoS₂, single Mo, single Tc or Nb & single S atoms respectively, N, N' & N'' respectively represent the number of Mo atom, number of S atom, and number of Tc or Nb atom present in impurities defected MoS₂ supercell structure, N''' denotes total number of atoms present in defected supercell structures.

Furthermore, water molecule is adsorbed on different positions at 2.45 Å distance above the surface of Tc-MoS₂ & Nb-MoS₂, it is found that adsorbed water molecule (i.e., physio-adsorption) is at 2.45 Å distance vertically from the center position of MoS₂ are more stable materials, which are shown in Fig. 1(c-d). This is because they have minimum ground state energies (-1810.08 Ry of Tc-W-MoS₂ & -1747.98 Ry of Nb-W-MoS₂ respectively), and maximum binding energies (7.89 eV of Tc-W-MoS₂ & 6.98 eV of Nb-W-MoS₂ respectively) than other positions of adsorbed water molecule on defected MoS₂. These binding energies are fairly agree with reported value of other stable 2D materials [52, 53]. We have also estimated the adsorption energy of water molecule with Tc-MoS₂ & Nb-MoS₂ by using equation (3), and found to be 2.17 eV & 2.08 eV respectively. They are comparable with the reported values of water adsorption on 2D materials [9, 25, 27]. Hence, water adsorbed defected materials are stable 2D materials.

$$E_a = E_{Tc/Nb-MoS_2} + E_{H_2O} - E_{Tc/Nb-W-MoS_2} \quad (3)$$

where, E_a is the adsorption energy of water molecule with Tc-MoS₂ or Nb-MoS₂ materials,

$E_{Tc/Nb-MoS_2}$ & E_{H_2O} , are the ground state energy of Tc-MoS₂ or Nb-MoS₂ and adsorbed water molecule respectively. From the estimation of binding energies of Tc-MoS₂, Nb-MoS₂, Tc-W-MoS₂ and Nb-W-MoS₂ materials, we concluded that Tc impurity defected MoS₂ are more compact than Nb impurity defected MoS₂ structure.

3.2 Electronic and Magnetic Properties

Solid materials are made up of a large number of atoms, each with discrete energy levels for their electrons. When two similar atoms are brought together, their atomic orbitals overlap and split into distinct energy levels for the molecule, forming a continuous range of energy known as an energy band. Understanding the band structure and density of states (DoS) of a solid material provide insight into its electronic properties. Magnetic properties of materials are explored by analyzing of their density of states (DoS) and partial density of states (PDOS). Therefore, electronic and magnetic properties of Tc & Nb impurities defected MoS₂ materials (Tc-MoS₂ & Nb-MoS₂), and water adsorbed on Tc & Nb impurities defected MoS₂ materials (Tc-W-MoS₂ & Nb-W-MoS₂) are studied by analyzing their band structure, DoS and PDOS plots. The band structures of Tc-MoS₂, Nb-MoS₂, Tc-W-MoS₂ and Nb-W-MoS₂ are shown in Fig. 2(a-d) respectively.

To understand the electronic properties of Tc-MoS₂, Nb-MoS₂, Tc-W-MoS₂ and Nb-W-MoS₂ materials, band structure and DoS calculations are performed. For that, we have chosen first Brillouin Zone of the reciprocal lattice. The x-axis of the band structure plot represents the high symmetric points Γ -M-K- Γ of the first Brillouin Zone, with the Γ -center being one of these points, and the y-axis corresponds to energy values. A total of 100 k-points are selected along these high symmetry points, which define a sampling path and irreducible region within the Brillouin Zone. In an ideal condition where the crystal is not under non-uniform strain or has imperfections. The repetition of the sampling path will cover the entire space of the Brillouin Zone. The band structure diagrams also have a horizontal dotted line, which represents the Fermi energy level. As seen in the band structure of Tc-MoS₂, band states of conduction band are moving closer to the Fermi energy level than that of valence band due to the rearrangement of spin states of electrons in the 3s, 3p orbitals of S atoms with 4d, 5s orbitals of Tc atom in structure. Hence, number of charge carriers in conduction band are greater than that in the valence band. It reveals that Tc-MoS₂ is a n-type semiconducting material having direct bandgap energy 1.01 eV at

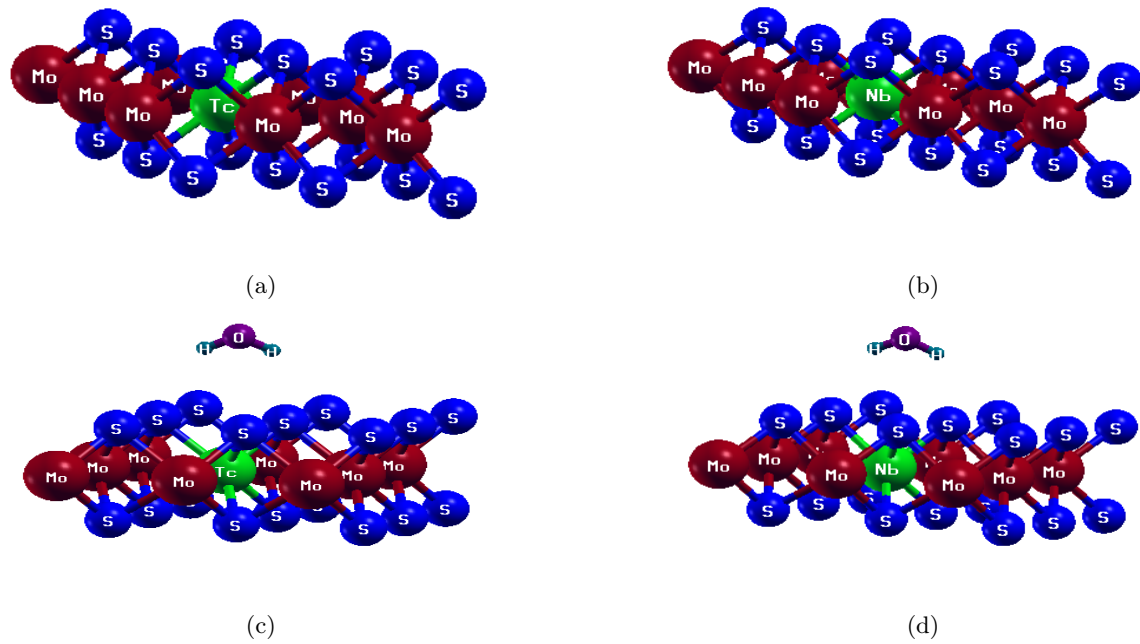


Figure 1: (Colour online) Supercell structure of (a) Tc impurity defect on MoS₂ (Tc-MoS₂), (b) Nb impurity defect on MoS₂ (Nb-MoS₂), (c) water adsorbed on Tc impurity defected MoS₂ (Tc-W-MoS₂), and (d) water adsorbed on Nb impurity defected MoS₂ (Nb-W-MoS₂).

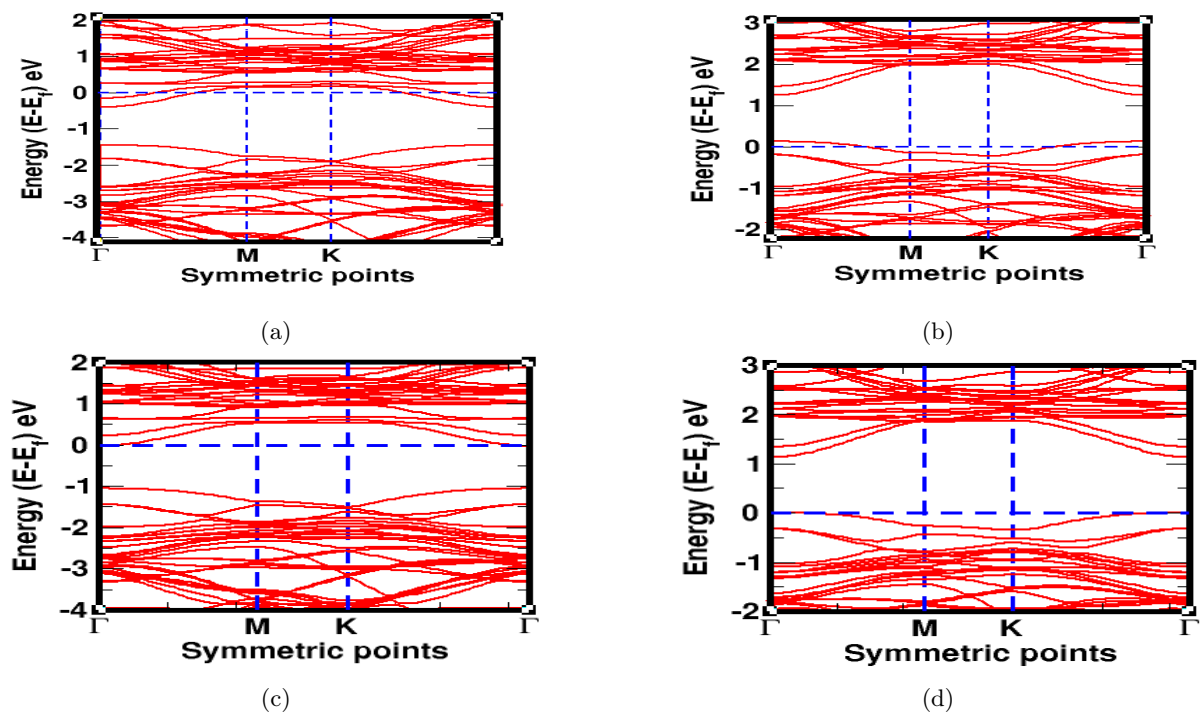


Figure 2: (Colour online) band structure plots of; (a) Tc-MoS₂, (b) Nb-MoS₂, (c) Tc-W-MoS₂ and (d) Nb-W-MoS₂. In figures, the horizontal dotted line represents the Fermi level, and the vertical dotted line intersects at high symmetry points.

Γ -point. This result is deviating from the properties of pristine MoS₂ [5–7], due to the presence of Tc impurity defects. We have estimated the bandgap energy of Nb-MoS₂ material by the analysis of its band structure calculations and found direct bandgap energy 1.10 eV in between the conduction band minima and valence band maxima at high symmetry Γ -point. Band states of valence band lies closer to the Fermi energy level than the band states of conduction band. This is because unpaired arrangement of spin states in the 3s, 3p orbitals of S atoms with 4d, 5s orbitals of Nb atom in structure. We concluded that Nb-MoS₂ is a p-type semiconductor material. This property is different than of pure MoS₂ [6,7], which is attributed due to the presence of Nb impurity defects in Nb-MoS₂. We have also analyzed the density of states (DoS) plots of Tc-MoS₂ & Nb-MoS₂ materials are shown in Fig. 3(a-b) respectively. It is found that distributed up-spin and down-spin states open bandgap energy due to the presence of unpaired spin states in the orbital of Tc & Nb atoms with S atoms respectively in structures. It also confirmed that Tc-MoS₂ is a n-type and Nb-MoS₂ is a p-type semiconductor material. From the analysis of above calculations, we concluded that impurity defects enhanced the electronic conductivity in materials. Semiconducting materials have wide applications due to their reliability, compactness and low cost; they are having probable future. Thus, p-type and n-type semiconductors find a wide variety of applications in the fields of memory, sensing, electronic, spintronic, optoelectronic devices [38, 42, 43].

Electronic properties of material are affected due to presence of moisture [25, 32]. So, we have investigated the electronic properties of water adsorbed on impurities defected MoS₂ through band structure and DoS plots. The band structures of Tc-W-MoS₂ & Nb-W-MoS₂ are shown in Fig. 2(c-d) respectively. In the band structure of Tc-W-MoS₂, a direct bandgap of 1.13 eV is observed at Γ -point. Its Fermi energy level being closer to the conduction band, which indicates that Tc-W-MoS₂ exhibits n-type semiconductor. The bandgap energy of Tc-W-MoS₂ has slightly greater value than that of Tc-MoS₂ material because adsorbed water molecule reduces binding energy (losses lubricity). It is observed that band states are slightly repelled towards conduction band from Fermi energy level in band structure is shown in Fig. 2(c). This is also confirmed by the analysis of its DoS plot is shown in Fig. 3(c). In DoS plot, bandgap energy is computed in between upper level of valence band and lower level of conduction band of distributed up- and down-spin states of electrons in the orbital of atoms present in the material. Furthermore, we have computed the bandgap energy of Nb-W-MoS₂ material on the basis of band structure and DoS

plots, are shown in Fig. 2(d) & 3(d) respectively. The band structure of Nb-W-MoS₂ reveals a direct band gap of 1.23 eV at Γ -point, which coupled with the Fermi energy being positioned near the valence band, indicates that Nb-W-MoS₂ has p-type of semiconducting properties. This obtained result is slightly deviated from the semiconducting behavior of pure MoS₂, which is due to the presence of adsorbed water molecule and Nb impurity defect on MoS₂. To confirm the predicted nature of Nb-W-MoS₂, we have analyzed the DoS of it and found that distributed up- and down-spin states are appeared near the valence band is shown in Fig. 3(d). This confirmed that it has p-type semiconducting properties. Water molecule can interact with impurities, altering their electronic and chemical states and subsequently modifying the band structure, and hence inflate the electronic properties than that of pristine MoS₂. Additionally, water adsorbing on the surface of MoS₂ can form a thin layer of water molecules and influence the electronic properties of the material. This can result in the creation of new energy levels leading to new electronic states and modifications to the bandgap energy.

Effect of impurities (Tc & Nb) defect on pristine MoS₂ and adsorbed water molecule on impurities defected MoS₂ initiate to switch in band structures and DoS plots. As a result, Fermi energy, Fermi shift, total ground state energy, bandgap energy, and binding energy of defected materials have changed than that of pristine form, they are presented in Table-1. The reason of shifting band states in band and DoS from pristine to defected systems are; H atom has one unpaired up-spin electron in 1s orbital, O atom has paired spins in 2s orbital and paired electrons in 2p_x sub-orbital, and unpaired electrons in 2p_y and 2p_z sub-orbitals. The Mo atom has one unpaired up-spin electron in 5s orbital and one in 4d orbital with sub-orbitals 4d_{xy}, 4d_{xz}, 4d_{yz}, 4d_{x²-y²}, 4d_{z²}. Tc atom has paired spins in 5s orbital and one unpaired up-spin electron in 4d_{xy}, 4d_{xz}, 4d_{yz}, 4d_{x²-y²} & 4d_{z²} sub-orbitals. Nb atom has one unpaired up-spin electron in 5s orbital, one unpaired electron in 4d_{xy}, 4d_{xz}, 4d_{yz} & 4d_{x²-y²} and one vacant in 4d_{z²} sub-orbital. The S atom has paired spins in 3p_x sub-orbital and one unpaired up-spin electron in 3p_y & 3p_z sub-orbitals. These different electronic configurations affect the electronic and magnetic properties of materials.

The magnetic properties of materials can be evaluated by analyzing their density of states (DoS) and partial density of states (PDOS). These plots provide insight into the distributions of electrons with up- and down-spins within the material. When the distributions are symmetrical, meaning that the number of up- and down-spins of electrons are equal, magnetic moments of the electrons canceled

out, resulting in non-magnetic properties. However, when the distributions are asymmetrical, there are more up- or down-spins of electrons, it leads to a net magnetic moment, resulting in magnetic properties. This occurs when electrons of the atoms are not paired, meaning the presence of unpaired up- and down-spins of electrons in the atomic orbitals create certain magnetic moment, leading to magnetic properties of the material. In other words, the DoS and PDoS calculations of spin-polarized systems reveal the distributions of electrons with opposite spins in the material. If the number of electrons with opposite spins are equal, the material is non-magnetic. But, if the distributions are asymmetrical, the material has magnetic properties. The reason is that electrons with opposite spins can pair up, which cancels out their magnetic moments. But, if there are more electrons with one spin direction, the material will have a net magnetic moment, making it magnetic. The PDoS plot of Tc-MoS₂, Nb-MoS₂, Tc-W-MoS₂ and Nb-W-MoS₂ are presented in Fig. 4(a-d) respectively.

DoS plots (Fig. 3(a-d)), and PDoS plots (Fig. 4(a-d)) come up with the information's of distributed spin states in electronic orbitals of atoms present in the materials. The DoS plots of considered materials give insight into how the impurities and adsorbed water molecule affect the electronic and magnetic properties of MoS₂. Hence, DoS used to understand how the defects and adsorbed water molecule influenced the electronic and magnetic properties of materials. The detail investigation of magnetic properties of Tc-MoS₂, Nb-MoS₂, Tc-W-MoS₂ and Nb-W-MoS₂ materials, we have analyzed PDoS plots. In PDoS analysis, we especially focused the impact of distributed electronic spin states in the orbitals of atoms present in the materials. It shows how the electrons are spread among different orbitals and how impuri-

ties and water exposure affect these distributions, providing a deeper understanding of how these factors affect the magnetic properties of materials. In DoS & PDoS plots of Tc-MoS₂, Nb-MoS₂, Tc-W-MoS₂ and Nb-W-MoS₂, we found that the distributions of up-spin and down-spin electronic states near the Fermi level are asymmetrical. We determined the net magnetic moment arising from up-spin and down-spin of electrons in each orbital of the constituent atoms. Magnetic moment in the 4p, 4d & 5s orbitals of Mo atom have 0.03 $\mu\text{B}/\text{cell}$, 0.11 $\mu\text{B}/\text{cell}$ & 0.00 $\mu\text{B}/\text{cell}$; 3s & 3p orbitals of S atoms have 0.00 $\mu\text{B}/\text{cell}$ & 0.05 $\mu\text{B}/\text{cell}$; and 4p, 4d & 5s orbitals of Tc atom has -0.01 $\mu\text{B}/\text{cell}$, -0.03 $\mu\text{B}/\text{cell}$ & 0.00 $\mu\text{B}/\text{cell}$ values in Tc-MoS₂ material. Total magnetic moment of Tc-MoS₂ is found to be 0.15 $\mu\text{B}/\text{cell}$, and the magnetic moment based on integrated density of states (IDoS) is also estimated to be 0.15 $\mu\text{B}/\text{cell}$. The magnetic moment generated in Tc-MoS₂ is mainly due to the 4d orbital of Mo, 3p orbital of S, and 4d orbital of Tc atoms. Thus, Tc-MoS₂ is considered a magnetic material. The magnetic properties of Nb-MoS₂ are investigated by the analysis of its PDoS plot. It is determined that the 4p, 4d & 5s orbitals of Mo atoms contributed 0.01 $\mu\text{B}/\text{cell}$, 0.09 $\mu\text{B}/\text{cell}$, & 0.00 $\mu\text{B}/\text{cell}$ magnetic moments respectively. Additionally, the magnetic moment carried out by 3s & 3p orbitals of S atoms have 0.00 $\mu\text{B}/\text{cell}$ & 0.01 $\mu\text{B}/\text{cell}$; 4p, 4d & 5s orbitals of Nb atoms have -0.02 $\mu\text{B}/\text{cell}$, -0.08 $\mu\text{B}/\text{cell}$ & 0.00 $\mu\text{B}/\text{cell}$ respectively. Total magnetic moment of Nb-MoS₂ is found to be 0.01 $\mu\text{B}/\text{cell}$. The same value of magnetic moment of Nb-MoS₂ is determined by integrated density of states (IDoS) calculation. The dominant contribution of magnetic moment by 4d orbital of Mo, 3p orbital of S, and 4d orbital of Nb atoms in Nb-MoS₂. Hence, Nb-MoS₂ is a weak magnetic material.

Table 1: Bandgap energy (E_g), Fermi energy (E_f), Fermi energy shift (E_s), total ground state energy (E_t), and binding energy (E_b) of MoS₂, Tc-MoS₂, Nb-MoS₂, Tc-W-MoS₂, and Nb-W-MoS₂ materials.

Materials	E_f (eV)	E_g (eV)	E_s (eV)	E_t (Ry)	E_b (eV)
MoS ₂	-1.89	1.80	-	-1741.61	6.83
Tc-MoS ₂	-0.83	1.01	1.06	-1774.07	7.32
Nb-MoS ₂	-2.26	1.10	-0.37	-1711.97	6.42
Tc-W-MoS ₂	-1.01	1.13	0.88	-1810.08	7.89
Nb-W-MoS ₂	-2.42	1.23	-0.53	-1747.98	6.98

Table 2: Total magnetic moment (μ) of Tc-MoS₂, Nb-MoS₂, Tc-W-MoS₂, and Nb-W-MoS₂ materials obtained by asymmetrically distributed up-spin and down-spin of electrons in various orbitals of different atoms.

Orbitals / Materials	Tc-MoS ₂	Nb-MoS ₂	Tc-W-MoS ₂	Nb-W-MoS ₂
μ of 4p-Mo	0.03	0.01	0.03	0.00
μ of 4d-Mo	0.11	0.09	0.12	0.08
μ of 5s-Mo	0.00	0.00	0.00	0.00
μ of 3s-S	0.00	0.00	0.00	0.00
μ of 3p-S	0.05	0.01	0.03	0.01
μ of 4p-Tc	-0.01	-	0.01	-
μ of 4d-Tc	-0.03	-	-0.04	-
μ of 5s-Tc	0.00	-	0.00	-
μ of 4p-Nb	-	-0.02	-	0.00
μ of 4d-Nb	-	-0.08	-	-0.07
μ of 5s-Nb	-	0.00	-	0.00
μ of 2s-O	-	-	0.00	0.00
μ of 2p-O	-	-	0.00	0.00
μ of 1s-H	-	-	0.00	0.00
Total (μ) μ_B /cell	0.15	0.01	0.15	0.02

Furthermore, we have estimated the magnetic properties of Tc-W-MoS₂ and Nb-W-MoS₂ materials by interpreting their PDoS plots. The magnetic properties of Tc-W-MoS₂ are predicted by the estimation of distributed electronic spin states in the individual orbital of each atom present in the structure. The 4p, 4d 5s orbitals of Mo atoms have magnetic moments 0.03 μ_B /cell, 0.12 μ_B /cell 0.00 μ_B /cell respectively in structure. Also, 3s 3p orbitals of S atoms have magnetic moments 0.00 μ_B /cell 0.03 μ_B /cell; and 4p, 4d 5s orbitals of Tc atom has magnetic moments 0.01 μ_B /cell, -0.04 μ_B /cell 0.00 μ_B /cell respectively. Similarly, we have estimated the magnetic moment generated by 2s 2p orbitals of O atom have values 0.00 μ_B /cell 0.00 μ_B /cell; and 1s orbital of H has value 0.00 μ_B /cell. The total magnetic moment of Tc-W-MoS₂ is obtained 0.15 μ_B /cell. We also have estimated the magnetic moment 0.15 μ_B /cell based on integrated density of states (IDoS). The magnetic moment in Tc-W-MoS₂ is mainly created by the dominant effect of 4d orbital of Mo, 3p orbital of S, and 4d orbital of Tc atoms. Therefore, Tc-W-MoS₂ is considered to be a magnetic material. In Nb-W-MoS₂, the magnetic moment is observed in 4p, 4d, 5s orbitals of Mo atoms have 0.00 μ_B /cell, 0.08 μ_B /cell 0.00 μ_B /cell respectively. The magnetic moments given by 3s 3p orbitals of S atoms have values 0.00 μ_B /cell 0.00 μ_B /cell respectively. The 4p, 4d 5s orbitals of Nb atom has magnetic moments of 0.00 μ_B /cell, -0.07 μ_B /cell 0.00 μ_B /cell; and 2s & 2p orbitals of O has magnetic moments of 0.00 μ_B /cell 0.00 μ_B /cell, while the 1s orbital of H atom has a magnetic moment of value 0.00 μ_B /cell. The total magnetic moment of Nb-W-MoS₂ is found to be 0.02 μ_B /cell. Magnetic moment of Nb-W-MoS₂

also estimated through integrated density of states (IDoS), and found to be 0.02 μ_B /cell. The main source of magnetic moment in Nb-W-MoS₂ is 4d orbitals of Mo, 4d orbitals of Nb, and 3p orbital of S atoms in the structure. Thus, Nb-W-MoS₂ is weak a magnetic material. From the above calculations of magnetic moment in the materials, we found that impurity atoms on MoS₂, and adsorbed water molecule on impurities defected MoS₂ intensified the magnetic properties in the materials. The detail calculations of magnetic moment of Tc-MoS₂, Nb-MoS₂, Tc-W-MoS₂ and Nb-W-MoS₂ materials by the analysis of individual electronic orbital of atoms are given in Table-2.

4 Conclusions

Structural, electronic, and magnetic properties of Tc-MoS₂, Nb-MoS₂, Tc-W-MoS₂ and Nb-W-MoS₂ materials have been explored by spin-polarized density functional theory (DFT) method with van der Waals (vdWs) corrections (DFT-D2) approach through computational tool Quantum ESPRESSO. It is found that Tc-MoS₂, Nb-MoS₂, Tc-W-MoS₂ and Nb-W-MoS₂ are stable materials. The nature of materials has been studied through their band structure and density of states (DoS) calculations, and the result showed that all materials have narrow bandgap semiconducting properties. The bandgap energy of Tc-MoS₂, Nb-MoS₂, Tc-W-MoS₂ and Nb-W-MoS₂ have values 1.01 eV, 1.10 eV, 1.15 eV and 1.23 eV respectively. The electronic band states of conduction band are appeared close to the Fermi energy level of Tc-MoS₂ Tc-W-MoS₂, and hence they are n-type semiconductor materials. On the other hand, band states of valence band are ap-

peared near the Fermi energy level than the band states of conduction band in band structure and DoS plots of Nb-MoS₂ Nb-W-MoS₂ materials. It reflects that they have p-type semiconducting properties. The presence of impurities in MoS₂ and adsorbed water molecule on impurities defected MoS₂ affects the semiconducting behavior of pure MoS₂ by shifting the band states with respect to Fermi energy level. Band structure calculations and DoS analysis revealed that the impact of water molecules on the surface of MoS₂ can alter its electronic properties, leading to the creation of new energy levels and modifications to the bandgap, resulting in new electronic states. Magnetic properties of considered materials are studied on the basis of their density of states (DoS) and partial density of states (PDoS) calculations, and found that all materials have magnetic properties. Magnetic moment of materials is developed due to presence of unpair electronic spin states in the individual orbitals of atoms present in the material. The estimated magnetic moment of Tc-MoS₂, Nb-MoS₂, Tc-W-MoS₂ and Nb-W-MoS₂ have values 0.15 μB /cell, 0.01 μB /cell, 0.15 μB /cell 0.02 μB /cell respectively. The significant values of magnetic moment are given in materials due to the presence of unpaired spin state in the 4d orbitals of Mo atoms, 3p orbital of S atoms, and 4d orbitals of Tc Nb atoms. Hence, electronic and magnetic properties are enhanced in the materials due to the presence of impurity atoms on MoS₂ and adsorbed water molecule on impurities defected MoS₂.

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